

SILICATES

- Tectosilicates

Name	Idealized composition	Functional importance
Quartz	SiO_2	Single most common mineral in soil, often accounting for 50% or more of the total dry weight. Normally sand and silt sized. No chemical charge, no shrink-swell. Surface area of 1 m ² /g
Opal	$\text{SiO}_2 \cdot 2\text{H}_2\text{O}$	Hydrated form of silica. Normally sand and especially silt sized. No chemical charge, no shrink-swell. Surface area of 1 m ² /g. Presence generally indicates either volcanic ash or plant-created phytoliths.
Orthoclase	KSi_3AlO_8	Weatherable feldspar that releases the macronutrient K ⁺ . Normally sand and silt sized. No chemical charge, no shrink-swell. Surface area of 1 m ² /g.
Anorthite	$\text{CaSi}_3\text{AlO}_8$	Weatherable feldspar that releases the macronutrient Ca ²⁺ . Normally sand and silt sized. No chemical charge, no shrink-swell. Surface area of 1 m ² /g.
Albite	$\text{NaSi}_3\text{Al}_2\text{O}_8$	Weatherable feldspar that releases Na ⁺ . Normally sand and silt sized. No chemical charge, no shrink-swell. Surface area of 1 m ² /g.

- Phyllosilicates

Name	Idealized composition	Functional importance
Muscovite	$\text{K}(\text{Si}_{3.0}\text{Al}_{1.0})(\text{Al}_{2.0})\text{O}_{10}(\text{OH})_2$	Weatherable dioctahedral mica that will convert into illite as it releases K ⁺ . Often silt and sand sized, occasionally clay sized. No shrink-swell. Low surface area 3 m ² /g
Biotite	$\text{K}(\text{Si}_{4.0})(\text{Mg}_{3.0})\text{O}_{10}(\text{OH})_2$	Weatherable trioctahedral mica releases K ⁺ . Often silt and sand sized, occasionally clay sized. No shrink-swell. Low surface area 3 m ² /g
Illite	$\text{K}_{0.8}(\text{Si}_{3.2}\text{Al}_{0.8})(\text{Al}_{2.0})\text{O}_{10}(\text{OH})_2$	Commonly coarse clay sized, especially coarse clay sized, with CEC in the range of 10 to 40 cmol _c kg ⁻¹ . No shrink-swell and moderate surface area (e.g., 50 m ² /g)
Vermiculite	$\text{X}_{0.7}(\text{Si}_{3.3}\text{Al}_{0.7})(\text{Al}_{2.0})\text{O}_{10}(\text{OH})_2$	Commonly clay sized with CEC in the range of 100 to 200 cmol _c kg ⁻¹ . Moderate shrink-swell and very high surface area (e.g., 700 m ² /g). Can have K ⁺ -fixation and/or NH ₄ ⁺ -fixation.
Smectite (Beidellite)	$\text{X}_{0.4}(\text{Si}_{3.6}\text{Al}_{0.4})(\text{Al}_{2.0})\text{O}_{10}(\text{OH})_2$	Commonly fine clay sized with CEC in the range of 60 to 140 cmol _c kg ⁻¹ . Huge shrink-swell and moderate to high surface area (e.g., 250 m ² /g). Layer charge is in the tetrahedral sheet.
Smectite (Montmorillonite)	$\text{X}_{0.4}(\text{Si}_4)(\text{Al}_{1.6}\text{Mg}_{0.4})\text{O}_{10}(\text{OH})_2$	Commonly fine clay sized with CEC in the range of 60 to 140 cmol _c kg ⁻¹ . Huge shrink-swell and moderate surface area (e.g., 250 m ² /g). Layer charge is in the octahedral sheet.
Chlorite	$\text{X}_{0.2}(\text{Si}_{3.6}\text{Al}_{0.2})(\text{Mg}_3)\text{O}_{10}(\text{OH})_2$	Commonly clay sized with CEC in the range of 10 to 40 cmol _c kg ⁻¹ . No shrink-swell. Moderate surface area (e.g., 30 m ² /g). Chlorite is a trioctahedral mineral akin to biotite.
Kaolinite	$\text{Si}_2\text{Al}_2\text{O}_5(\text{OH})_4$	Commonly clay sized with CEC in the range of 1 to 10 cmol _c kg ⁻¹ . No shrink-swell. Moderate surface area (e.g., 30 m ² /g).

CARBONATES, SULFATES, & PHOSPHATES

Name	Idealized composition	Functional importance
Calcite	CaCO ₃	Commonly sand and silt sized soluble mineral that readily releases the Ca ²⁺ . Ground calcite (aka, "ag lime") is routinely added to acidic soils in order to raise the soil pH.
Dolomite	CaMg(CO ₃) ₂	Commonly sand and silt sized soluble mineral that readily releases the Ca ²⁺ and Mg ²⁺ . Ground dolomite (aka, "dolomitic lime") is routinely added to acidic soils in order to raise the soil pH.
Siderite	FeCO ₃	Commonly sand and silt sized soluble mineral that readily releases the Fe ²⁺ or Fe ³⁺ depending on redox conditions in the soil.
Gypsum	CaSO ₄ *nH ₂ O	Highly soluble sand and silt sized mineral that readily releases two critical plant nutrients, Ca ²⁺ and SO ₄ ²⁻ . No shrink-swell. Little surface area.
Apatite	Ca ₅ (PO ₄) ₃ (F,Cl)	Low solubility sand and silt sized mineral that is the only natural occurring mineral source of PO ₄ ³⁻ in soils. No shrink-swell. Little surface area.

OXIDES & HYDROXIDES

Name	Idealized composition	Functional importance
Goethite	FeO(OH)	Goethite is the most common oxide mineral in soils. It is frequently a weathering product formed when free Fe ⁺² precipitates into Fe ³⁺ during a soil oxidation reaction. Goethite can have pH-dependent CEC and AEC in the range of 0 to 5 cmol _c kg ⁻¹ . No shrink-swell. Moderate to high surface area (e.g., 60 to 200 m ² /g).
Hematite	Fe ₂ O ₃	Hematite is a common oxide mineral in soils that forms during organic matter decomposition. Hematite can have pH-dependent CEC and AEC in the range of 0 to 5 cmol _c kg ⁻¹ . No shrink-swell. Moderate to high surface area (e.g., 100 to 200 m ² /g).
Ferrihydrite	Fe ₅ O ₈ H ⁺ *nH ₂ O	Ferrihydrite is a poorly crystallized iron oxide that has poorly understood but significant ion exchange and surface area. No shrink-swell.
Corundum	Al ₂ O ₃	Corundum is entirely inherited from parent materials and generally considered pedogenically non-reactive. No CEC, no shrink-swell, low surface area.
Gibbsite	Al(OH) ₃	Gibbsite accumulates in soils that have weathering of aluminosilicates and extremely high leaching losses – i.e., the Al doesn't leave the soil even as other cations do. Gibbsite can have pH-dependent CEC and AEC in the range of 0 to 10 cmol _c kg ⁻¹ . No shrink-swell. Low surface area (e.g., 5 to 10 m ² /g).